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Europäisches Patentamt  
European Patent Office  
Office européen des brevets



(11) Publication number:

**0 435 253 A1**

(12)

## EUROPEAN PATENT APPLICATION

(21) Application number: 90125500.0

(22) Date of filing: 27.12.90

(51) Int. Cl.<sup>5</sup>: **C10M 105/36**, C10M 105/38,  
C10M 105/42, C10M 111/04,  
C10M 169/04, C09K 5/04,  
//(C10M111/04,105:36,105:38,  
105:42,107:34),(C10M169/04,  
105:36,105:38,105:42,107:34,  
129:18,137:02),C10N40:30

(30) Priority: 28.12.89 JP 341244/89  
28.12.89 JP 341245/89  
20.04.90 JP 105772/90  
14.05.90 JP 121133/90

(43) Date of publication of application:  
03.07.91 Bulletin 91/27

(84) Designated Contracting States:  
DE DK GB IT

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(54) Refrigerator oils for use with hydrogen-containing halogenocarbon refrigerants.

(57) A refrigerator oil for use with a hydrogen-containing halogenocarbon refrigerant, comprising at least one kind of an ester selected from the group consisting of a specific pentaerythritol ester such as an ester of pentaerythritol with a mono- or dicarboxylic acid, a specific polyol ester such as an ester of trimethylolethane with a mono- or dicarboxylic acid, a specific ester such as an ester of ethylene glycol and a dicarboxylic acid, and a specific polyol ester synthesized from a neopentyl type polyhydric alcohol, a monocarboxylic acid and a dicarboxylic acid.

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## REFRIGERATOR OILS FOR USE WITH HYDROGEN-CONTAINING HALOGENOCARBON REFRIGERANTS

Background of the Invention

## 1. Field of the Invention

5 This invention relates to a lubricating oil for compressors of refrigerators using therein a hydrogen-containing halogenocarbon as a refrigerant (the oil being hereinafter referred to as "a refrigerator oil for use with a hydrogen-containing halogenocarbon refrigerant") and, more specifically, it relates to such a refrigerator oil which comprises a specific ester as a base oil and is superior in various properties.

## 10 2. Prior Art

Generally, naphthenic mineral oils, paraffinic mineral oils, alkylbenzenes, polyglycolic oils, ester oils and mixtures thereof, which have each a kinematic viscosity of 10 - 200 cSt at 40° C, as well as said oils incorporated with suitable additives have been used as refrigerator oils.

15 On the other hand, chlorofluorocarbons (CFCs) type refrigerants, such as CFC-11, CFC-12, CFC-113 and HCFC-22, have been used for refrigerators.

Of these CFCs, CFCs such as CFC-11, CFC-12 and CFC-113, which are obtained by substituting all the hydrogen atoms of hydrocarbons thereof by halogen atoms including chlorine atoms, may lead to the destruction of the ozone layer, and therefore, the use of the CFCs has been controlled. Accordingly, 20 hydrogen-containing halogenocarbons, such as HFC-134a and HFC-152a, have been being used as substitutes for CFCs. HFC-134a is especially hopeful as a substitute refrigerant since it is similar in thermodynamic properties to CFC-12 which has heretofore been used in many kinds of refrigerators of home cold-storage chests, air-conditioners and the like.

Refrigerator oils require various properties, among which their compatibility with refrigerants is extremely important in regard to lubricity and system efficiency in refrigerators. However, conventional refrigerator oils comprising, as the base oils, naphthenic oils, paraffinic oils, alkylbenzenes, heretofore known ester oils and the like, are hardly compatible with hydrogen-containing halogenocarbons such as HFC-134a. Therefore, if said conventional refrigerator oils are used in combination with HFC-134a, the resulting mixture will separate into two layers at normal temperature thereby to degrade the oil-returnability 30 which is the most important within the refrigeration system and cause various troubles such as a decrease in refrigeration efficiency, the deterioration of lubricity and the consequent seizure of the compressor within the system whereby the refrigerator oils are made unsuitable for use as such. In addition, polyglycolic oils are also known as refrigerator oils for their high viscosity index and are disclosed in, for example, Japanese Pat. Gazettes Nos. Sho. 57-42119 and Sho. 61-52880 and Japanese Pat. Appln. Laid-Open Gazette No. 35 Sho. 57-51795. However, the polyglycolic oils concretely disclosed in these prior art publications are not fully compatible with HFC-134a thereby raising the same problems as above and rendering them unusable.

Further, U.S. Patent No. 4,755,316 discloses polyglycolic refrigerator oils which are compatible with HFC-134a and U.S. Patent No. 4,851,144 discloses refrigerator oils comprising a mixture of an ester and a polyglycol which are compatible with HFC-134a. In addition, the present inventors developed polyglycolic 40 refrigerator oils which have excellent compatibility with HFC-134a as compared with conventional known refrigerator oils, filed an application for a patent for the thus developed polyglycolic refrigerator oils and have already obtained a patent (U.S.P. No. 4,948,525) therefor. It has been found, however, that the polyglycolic oils raise problems as to their high compatibility with water and inferior electrical insulating property.

45 On the other hand, refrigerator oils used in compressors of home refrigerators and the like are required to have a high electrical insulating property. Among the known refrigerator oils, alkylbenzenes and the mineral oils have the highest insulating property, but they are hardly compatible with hydrogen-containing halogenocarbons such as HFC-134a as mentioned above. Therefore, there had been developed no refrigerator oils having both high compatibility with hydrogen-containing halogenocarbons such as HFC- 50 134a and a high insulating property before the accomplishment of the present invention.

Summary of the Invention

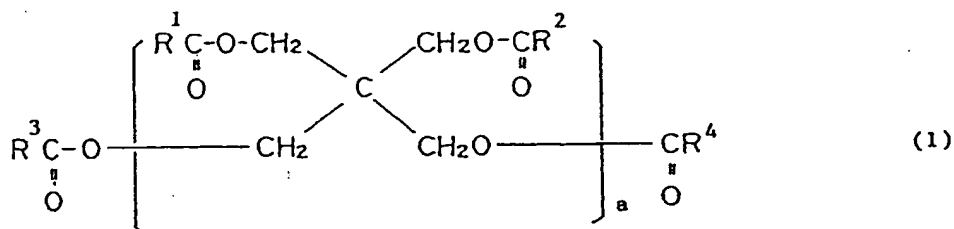
The present inventors made various intensive studies in attempts to develop refrigerator oils which can meet the aforesaid requirements and, as the result of their studies, they found that esters having specific

structures have excellent compatibility with hydrogen-containing halogenocarbons such as HFC-134a, and a high electrical insulating property as well as excellent lubricity. This invention is based on this finding.

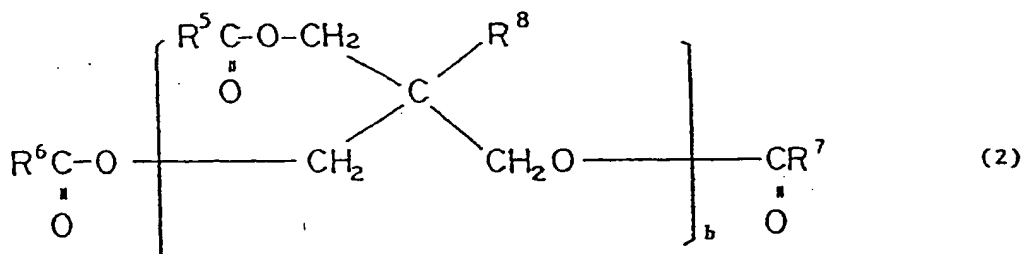
The object of this invention is to provide refrigerator oils for use with hydrogen-containing halogenocarbons refrigerants, the oils comprising as a major component (or a base oil) at least one kind of ester having a specific structure and having excellent compatibility with hydrogen-containing halogenocarbons such as HFC-134a, and a high electrical insulating property.

The refrigerator oil of the present invention is characterized in that it comprises as a base oil at least one kind of an ester selected from the group consisting of:

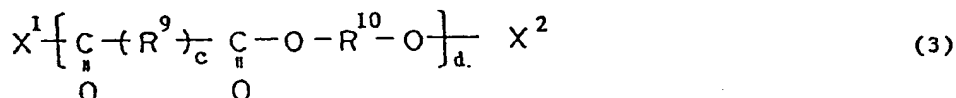
[I] a pentaerythritol ester represented by the general formula (1)



[III] a polyol ester represented by the general formula (2)



[III] an ester represented by the general formula (3)



45 and

[IV] a polyol ester prepared by the synthesis of, as raw materials, (a) a neopentyl type polyhydric alcohol having 5-6 carbon atoms and 3-4 hydroxyl groups, (b) a monobasic carboxylic acid and (c) a dibasic carboxylic acid.

First, the pentaerythritol esters [I] will be explained in detail. In the formula (1), R<sup>1</sup>-R<sup>4</sup> may be identical with, or different from, each other and are each a group selected from the group consisting of straight-chain alkyl groups having 3-11 carbon atoms, preferably 3-7 carbon atoms, branched-chain alkyl groups having 3-15 carbon atoms, preferably 4-11 carbon atoms and cycloalkyl groups having 6-12 carbon atoms, preferably 6-8 carbon atoms. The cycloalkyl groups in this invention may include alkylcycloalkyl groups. In addition, a ratio of the straight-chain alkyl groups to the total alkyl groups is not more than 60%, preferably not more than 50%. Further, a is an integer of 1-3. Thus, the formula(1) illustrates monopentaerythritol esters, dipentaerythritol esters and tripentaerythritol esters.

R<sup>1</sup>-R<sup>4</sup> are each exemplified by n-propyl group, n-butyl group, n-pentyl group, n-hexyl group, n-heptyl group, n-octyl group, n-nonyl group, n-decyl group, n-undecyl group, iso-propyl group, iso-butyl group, iso-

pentyl group, iso-hexyl group, iso-heptyl group, iso-octyl group, iso-nonyl group, iso-decyl group, iso-undecyl group, iso-dodecyl group, iso-tridecyl group, iso-tetradecyl group, iso-pentadecyl group, cyclohexyl group, cycloheptyl group, cyclooctyl group, cyclononyl group, cyclodecyl group, cycloundecyl group, cyclododecyl group, methylcyclohexyl group, ethylcyclohexyl group, propylcyclohexyl group, butylcyclohexyl group, pentylcyclohexyl group or hexylcyclohexyl group.

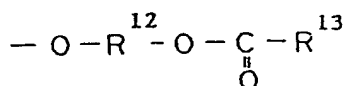
The pentaerythritol esters [I] are esters of pentaerythritol, dipentaerythritol or tripentaerythritol and a monocarboxylic acid, and are ordinarily obtained by the reaction of pentaerythritol, dipentaerythritol, tripentaerythritol or a mixture thereof with a mixture of at least one carboxylic acid having the aforementioned alkyl group.

Secondly, the polyol esters [II] will be explained in detail. In the formula (2),  $R^5$ - $R^7$  may be identical with, or different from, each other and are each a group selected from the group consisting of straight-chain alkyl groups having 3-11 carbon atoms, preferably 3-7 carbon atoms, branched-chain alkyl groups having 3-15 carbon atoms, preferably 4-11 carbon atoms and cycloalkyl groups having 6-12 carbon atoms, preferably 6-8 carbon atoms. The cycloalkyl groups in this invention may include alkylcycloalkyl groups. With respect to the above  $R^5$ - $R^7$ , a ratio of the straight-chain alkyl groups to the total alkyl groups (including cycloalkyl groups) is not more than 60%, preferably not more than 50%. In addition,  $R^8$  is a group selected from the group consisting of methyl group, ethyl group and propyl group. Further, b is an integer of 1-3. Thus, the formula (2) illustrates trimethylolethane esters, trimethylolpropane esters, trimethylolbutane esters and dimers-trimers thereof.

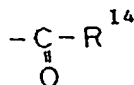
$R^5$ - $R^7$  are each exemplified by n-propyl group, n-butyl group, n-pentyl group, n-hexyl group, n-heptyl group, n-octyl group, n-nonyl group, n-decyl group, n-undecyl group, iso-propyl group, iso-butyl group, iso-pentyl group, iso-hexyl group, iso-heptyl group, iso-octyl group, iso-nonyl group, iso-decyl group, iso-undecyl group, iso-dodecyl group, iso-tridecyl group, iso-tetradecyl group, iso-pentadecyl group, cyclohexyl group, cycloheptyl group, cyclooctyl group, cyclononyl group, cyclodecyl group, cycloundecyl group, cyclododecyl group, methylcyclohexyl group, ethylcyclohexyl group, propylcyclohexyl group, butylcyclohexyl group, pentylcyclohexyl group or hexylcyclohexyl group.

The polyol esters [II] are esters of trimethylolethane, trimethylolpropane, trimethylolbutane or a dimer or trimer thereof and a monocarboxylic acid, and are ordinarily obtained by the reaction of trimethylolethane, trimethylolpropane, trimethylolbutane, a dimer or trimer thereof or a mixture thereof with a mixture of at least one carboxylic acid having the aforementioned alkyl group.

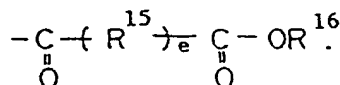
Thirdly, the esters [III] will be explained in detail. In the formula (3),  $X^1$  is a group represented by the general formula  $-OR^{11}$  or



and  $X^2$  is a group represented by the general formula



or



In addition,  $R^9$  and  $R^{15}$  are each a divalent hydrocarbon group having 1-8 carbon atoms, preferably 1-6 carbon atoms,  $R^{10}$  and  $R^{12}$  are each a divalent saturated hydrocarbon group having 2-16 carbon atoms, preferably 2-9 carbon atoms,  $R^{11}$  and  $R^{16}$  are each an alkyl group having 1-15 carbon atoms, preferably 1-12 carbon atoms, and  $R^{13}$  and  $R^{14}$  are each an alkyl group having 1-14 carbon atoms, preferably 1-11 carbon atoms. Further, c and e are each an integer of 0 or 1 and d is an integer of 0-30, preferably 1-30.

$R^9$  and  $R^{15}$  are each exemplified by a methylene group, ethylene group, propylene group, trimethylene

group, butylene group, tetramethylene group, pentamethylene group, hexamethylene group, heptamethylene group, octamethylene group, phenylene group, other unsaturated hydrocarbon groups or the like. R<sup>10</sup> and R<sup>12</sup> are each illustrated by nonamethylene group, decamethylene group, undecamethylene group, dodecamethylene group, tridecamethylene group, tetradecamethylene group, pentadecamethylene group, hexadecamethylene group, cyclohexylene group or the like in addition to the above alkylene groups (except for methylene group). R<sup>13</sup> and R<sup>14</sup> are each concretely exemplified by straight- or branched-chain alkyl groups, such as methyl group, ethyl group, propyl group, butyl group, pentyl group, hexyl group, heptyl group, octyl group, nonyl group, decyl group, undecyl group, dodecyl group, tridecyl group or tetradecyl group. R<sup>11</sup> and R<sup>16</sup> are each illustrated by a straight- or branched-chain alkyl group, such as a pentadecyl group, in addition to any one of the above alkyl groups.

A method for producing the esters [III] is not particularly limited. For example, a mixture of the esters [III] can be produced at one time by the esterification reaction of (i) a diol having none of ether linkages in the branched-chains thereof with (ii) a dicarboxylic acid and (iii) a monocarboxylic acid and/or (iv) a monohydric alcohol. The esters [III] are also produced by the reaction of the diol (i) with the monocarboxylic acid (iii) or the reaction of the dicarboxylic acid (ii) with the monohydric alcohol (iii).

The diols (i) used herein are those having 2-16 carbon atoms and are exemplified by alkylene glycols such as ethylene glycol, propylene glycol, butylene glycol, trimethylene glycol, tetramethylene glycol, pentamethylene glycol, hexamethylene glycol, heptamethylene glycol, octamethylene glycol, nonamethylene glycol, decamethylene glycol, neopentyl glycol, 2-ethyl-2-methyl-1,3-propanediol, 2-methyl-2-propyl-1,3-propanediol, 2,2-dimethyl-1,3-propanediol, 2-butyl-2-ethyl-1,3-propanediol, 3-methyl-1,5-pentanediol, 2,4-pentanediol, 2-methyl-2,4-pentanediol, 2-methyl-1,6-hexane, 1,4-cyclohexane dimethanol and 2,2-bis (4-hydroxycyclohexyl) propane.

The dicarboxylic acids (ii) used herein are those having 2-10 carbon atoms and are exemplified by saturated aliphatic dicarboxylic acids such as oxalic acid, malonic acid, succinic acid, glutaric acid, adipic acid, pimelic acid, suberic acid, azelaic acid, sebacic acid, methylmalonic acid, ethylmalonic acid, dimethylmalonic acid, methylsuccinic acid, 2,2-dimethylsuccinic acid, 2,3-dimethylsuccinic acid, 2-ethyl-2-methylsuccinic acid, 2-methylglutaric acid, 3-methylglutaric acid, 3,3-dimethylglutaric acid and 3-methyladipic acid; unsaturated aliphatic dicarboxylic acids such as maleic acid, fumaric acid, itaconic acid, citraconic acid and mesaconic acid; and aromatic dicarboxylic acids such as phthalic acid, isophthalic acid and terephthalic acid.

The monocarboxylic acids (iii) used herein are those having 2-15 carbon atoms and are exemplified by acetic acid, propionic acid, butanoic acid, pentanoic acid, hexanoic acid, heptanoic acid, octanoic acid, nonanoic acid, decanoic acid, undecanoic acid, dodecanoic acid, tridecanoic acid, tetradecanoic acid, pentadecanoic acid, 3-methylbutanoic acid, 2-methylbutanoic acid, 2-ethylhexanoic acid, 2,4-dimethylpentanoic acid, 3,3,5-trimethylhexanoic acid and benzoic acid.

The monohydric alcohols (iv) used herein are those having 1-15 carbon atoms and are exemplified by methanol, ethanol, propanol, butanol, pentanol, hexanol, heptanol, octanol, nonanol, decanol, undecanol, dodecanol, tridecanol, tetradecanol, pentadecanol, isopropanol, isobutanol, 2-methyl-1-butanol, 2,2-dimethyl-1-propanol, 3,3-dimethyl-1-butanol, 2-methyl-1-pentanol, 3-methyl-1-pentanol, 2,4,4-trimethyl-1-pentanol, 2,2,4-trimethyl-1-pentanol, 2-ethyl-4-methyl-1-pentanol and 2-ethyl-1-hexanol.

The molecular weight of the ester [III] is not particularly limited, but the number average molecular weight of the ester [III] is in the range of preferably 200-3000, more preferably 300-2000, to improve the compressor in sealability.

Fourthly, the polyol ester [IV] will be explained in detail.

The neopentyl type polyhydric alcohols (a) having 5-6 carbon atoms and 3-4 hydroxyl groups are illustrated by trimethylolethane, trimethylolpropane and pentaerythritol.

The monocarboxylic acids (b) preferably used herein are those having 2-15 carbon atoms and are exemplified by acetic acid, propionic acid, butanoic acid, pentanoic acid, hexanoic acid, heptanoic acid, octanoic acid, nonanoic acid, decanoic acid, undecanoic acid, dodecanoic acid, tridecanoic acid, tetradecanoic acid and pentadecanoic acid.

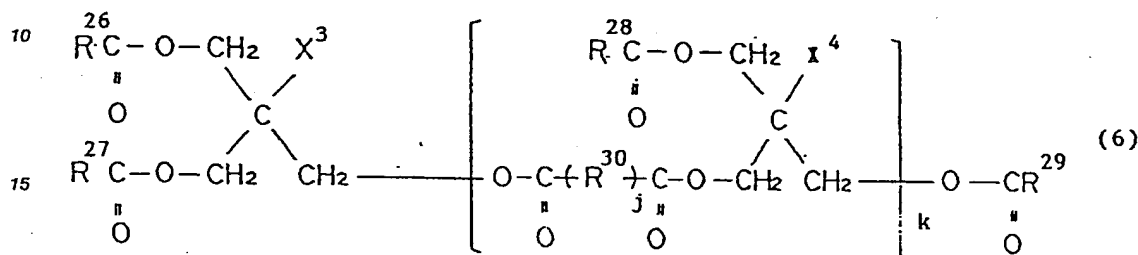
These carboxylic acids may be either straight-chain or branched-chain ones, but it is preferable to use the latter having a branched-chain structure in an amount of not less than 50 mol%, more preferably not less than 60 mol%.

The dicarboxylic acids (c) used herein are those having 2-10 carbon atoms and are exemplified by saturated aliphatic dicarboxylic acids such as oxalic acid, malonic acid, succinic acid, glutaric acid, adipic acid, pimelic acid, suberic acid, azelaic acid, sebacic acid, methylmalonic acid, ethylmalonic acid, dimethylmalonic acid, methylsuccinic acid, 2,2-dimethylsuccinic acid, 2,3-dimethylsuccinic acid, 2-ethyl-2-methylsuccinic acid, 2-methylglutaric acid, 3-methylglutaric acid and 3-methyladipic acid; unsaturated aliphatic

dicarboxylic acids such as maleic acid, fumaric acid, itaconic acid, citraconic acid and mesaconic acid; and aromatic dicarboxylic acids such as phthalic acid, terephthalic acid and isophthalic acid.

The esters [IV] are produced by the esterification reaction of the aforesaid component (a) with the component (b) and the component (c) under ordinary esterification conditions, for example, in the presence of an acid catalyst such as sulfuric acid and at a temperature of 100-180°C.

The chemical structure of the ester [IV] according to the present invention is illustrated by the following general formula (6)



wherein  $\text{X}^3$  and  $\text{X}^4$  may be identical with, or different from, each other and are each an alkyl group having 1-4 carbon atoms or a group represented by the general formula



$\text{R}^{26}$ -,  $\text{R}^{29}$  and  $\text{R}^{31}$  may be identical with, or different from each other and are each an alkyl group having 4-15 carbon atoms;  $\text{R}^{30}$  is an alkylene group having 1-8 carbon atoms;  $j$  is an integer of 0 or 1; and  $k$  is an integer of 1-5.

The molecular weight of the polyol ester [IV] is not particularly limited, but the number average molecular weight of the ester [IV] is in the range of preferably 200-3000, more preferably 300-2000, to improve the compressor in sealability.

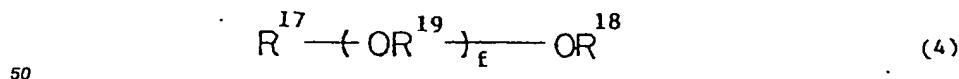
The products obtained by the methods as mentioned above may be refined to remove the by-products and/or unreacted reactants, but the by-products and/or unreacted reactants may be present in small amounts in the refrigerator oils of the present invention as far as they do not impair the excellent performances thereof.

In the preparation of the refrigerator oils of the present invention, the esters [I]-[IV] mentioned above may be used singly, or jointly as a mixture of at least two kinds of the esters.

The kinematic viscosities of the esters according to the present invention are in the range of preferably 2-150 cSt, more preferably 5-100 cSt at 100°C.

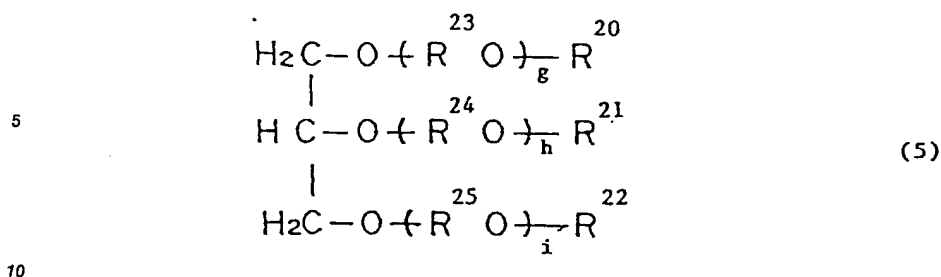
The refrigerator oil of the present invention may comprise as the only base oil at least one member selected from the above esters [I]-[IV] and, as required, it may additionally comprise other base oils for refrigerator oils. Among the other base oils, preferable ones are illustrated as follows:

a polyoxyalkylene glycol or an ether thereof represented by the general formula (4)



wherein  $\text{R}^{17}$  and  $\text{R}^{18}$  are each a hydrogen atom or an alkyl group having 1-18 carbon atoms,  $\text{R}^{19}$  is an alkylene group having 2-4 carbon atoms and  $f$  is an integer of 5-70, and

a polyoxyalkylene glycol glycerol ether represented by the general formula (5)



wherein  $\text{R}^{20}-\text{R}^{22}$  are each a hydrogen atom or an alkyl group having 1-18 carbon atoms,  $\text{R}^{23}-\text{R}^{25}$  are each an alkylene group having 2-4 carbon atoms and  $g-i$  are each an integer of 5-7.

These conventional oils represented by the formulae (4) and (5) may be used singly or jointly for adding to the refrigerator oil of this invention. Further, the refrigerator oil of this invention may be incorporated with paraffinic mineral oils, naphthenic mineral oils, poly $\alpha$ -olefins, alkylbenzenes and the like, but, in this case, the resulting mixed oil will be lowered in compatibility with hydrogen-containing halogonocarbons.

The amount of these conventional base oils so incorporated is not particularly limited as far as the excellent performances of the refrigerator oil of this invention are not impaired, but the esters [I]-[IV] should be present in the resulting mixed oil in a ratio of usually more than 50 % by weight, preferably not less than 70 % by weight of the total amount of the mixed oil.

The refrigerator oil composition according to this invention may be incorporated further with at least one kind of a phosphorous compound selected from the group consisting of phosphoric esters, acid phosphoric esters, amine salts of acid phosphoric esters, chlorinated phosphoric esters and phosphorous esters, to improve the oil composition in wear resistance and load resistance. These phosphorous compounds are esters of phosphoric acid or phosphorous acid and an alkanol or a polyether type alcohol, or derivatives of the esters. The phosphoric esters are exemplified by tributyl phosphate, triphenyl phosphate and tricresyl phosphate. The acid phosphoric esters are exemplified by ditetradecyl acid phosphate, dipentadecyl acid phosphate, dihexadecyl acid phosphate, diheptadecyl acid phosphate and dioctadecyl acid phosphate. The amine salts of acid phosphoric esters are exemplified by salts of the above acid phosphoric esters and amines such as methylamine, ethylamine, propylamine, butylamine, pentylamine, hexylamine, heptylamine, octylamine, dimethylamine, diethylamine, dipropylamine, dibutylamine, dipentylamine, dihexylamine, diheptylamine, dioctylamine, trimethylamine, triethylamine, tripropylamine, tributylamine, tripentylamine, trihexylamine, triheptylamine and trioctylamine. The chlorinated phosphoric esters are exemplified by tris-dichloropropyl phosphate, tris-chloroethyl phosphate, polyoxyalkylene-bis[di(chloroalkyl)] phosphate and tris-chlorophenyl phosphate. The phosphorous esters are exemplified by dibutyl phosphite, tributyl phosphite, dipentyl phosphite, tripentyl phosphite, dihexyl phosphite, trihexyl phosphite, diheptyl phosphite, triheptyl phosphite, dioctyl phosphite, trioctyl phosphite, dinonyl phosphite, didodecyl phosphite, diundecyl phosphite, triundecyl phosphite, didodecyl phosphite, tridodecyl phosphite, diphenyl phosphite, triphenyl phosphite, dicresyl phosphite, tricresyl phosphite and mixtures thereof. These phosphorous compounds may be added to the refrigerator oil in a ratio of 0.1-5.0 % by weight, preferably 0.2-2.0 % by weight, of the total amount of the refrigerator oil.

To further improve the refrigerator oil of the present invention in stability, it may be incorporated with at least one kind of an epoxy compound selected from the group consisting of phenylglycidyl ether type epoxy compounds, glycidyl ester type epoxy compounds, epoxidized fatty acid monoesters and epoxidized vegetable oils. The said phenylglycidyl ether type epoxy compounds used herein include phenylglycidyl ether and alkylphenylglycidyl ethers. The said alkylphenylglycidyl ethers are those having 1 to 3 alkyl groups having 1 to 13 carbon atoms, among which are preferred those having an alkyl group having 4 to 10 carbon atoms, such as butylphenylglycidyl ether, pentylphenylglycidyl ether, hexylphenylglycidyl ether, heptylphenylglycidyl ether, octylphenylglycidyl ether, nonylphenylglycidyl ether and decylphenylglycidyl ether. The said glycidyl ester type epoxy compounds include phenylglycidyl esters, alkylglycidyl esters and alkenylglycidyl esters with glycidyl benzoate, glycidyl acrylate, glycidyl methacrylate and the like being preferred.

The epoxidized fatty acid monoesters include esters of an epoxidized fatty acid having 12 to 20 carbon atoms and an alcohol having 1 to 8 carbon atoms, phenol or an alkylphenol. In particular, butyl, hexyl, benzyl, cyclohexyl, methoxyethyl, octyl, phenyl or butylphenyl esters of epoxidized stearic acid may preferably be used.



The epoxidized vegetable oils include epoxidized compounds of vegetable oils such as soybean oil, linseed oil and cottonseed oil.

Among these epoxy compounds, the preferable ones include phenylglycidyl ether type epoxy compounds and epoxidized fatty acid monoesters with the former being more preferable. The most preferred are phenylglycidyl ether, butylphenylglycidyl ether and mixtures thereof.

In a case where these epoxy compounds are to be incorporated in the refrigerator oil of the present invention, it is desirable that they be incorporated therein in a ratio of 0.1-5.0 % by weight, preferably 0.2-2.0 % by weight, of the total amount of the refrigerator oil.

Of course, both of the aforementioned phosphorus compounds and epoxy compounds may be used jointly.

To further enhance the refrigerator oil of this invention in performances, the refrigerator oil may be incorporated, as required, with heretofore known additives for a refrigerator oil, which include phenol-type antioxidants such as di-tert-butyl-p-cresol and bisphenol A; amine-type antioxidants such as phenyl- $\alpha$ -naphthylamine and N,N-di(2-naphthyl)-p-phenylenediamine; wear resistant additives such as zinc dithiophosphate; extreme pressure agents such as chlorinated paraffin and sulfur compounds; oiliness improvers such as fatty acids; antifoaming agents such as silicone-type ones; and metal inactivators such as benzotriazole. These additives may be used singly or jointly. The total amount of these additives added is ordinarily not more than 10 % by weight, preferably not more than 5 % by weight, of the total amount of the refrigerator oil.

The refrigerator oils of this invention comprising at least one of the esters [I]-[IV] as the base oil should have such viscosity and pour point as those which are normally suitable for an ordinary refrigerator oil, but they should desirably have a pour point of not higher than  $-10^{\circ}\text{C}$ , preferably  $-20^{\circ}\text{C}$  to  $-80^{\circ}\text{C}$ , to prevent them from solidification at a low temperature. Further, they should desirably have a kinematic viscosity of not less than 2cSt, preferably not less than 3cSt at  $100^{\circ}\text{C}$ , to keep the sealability of the compressor when used, while they should desirably have a kinematic viscosity of not more than 150cSt, preferably not more than 100cSt at  $100^{\circ}\text{C}$ , in view of their fluidity at a low temperature and the efficiency of heat exchange in the evaporator when used.

The refrigerants which may be used in refrigerators in which the lubricating oils (refrigerator oils) of the present invention are suitably used, include hydrogen-containing halogenocarbons such as trifluoromethane (HFC-23), pentafluoroethane (HFC-125), 1,1,2,2-tetrafluoroethane (HFC-134), 1,1,1,2-tetrafluoroethane (HFC-134a), 1,1-difluoroethane (HFC-152a), monochlorodifluoromethane (HCFC-22), 1-chloro-1,1-difluoroethane (HCFC-142b), dichlorotrifluoroethane (HCFC-123), monochlorotetrafluoroethane (HCFC-124), and mixtures thereof. Among these hydrogen-containing halogenocarbons, the chlorine-free type halogenocarbons such as HFC-23, HFC-125, HFC-134, HFC-134a and HFC-152a, are preferable with HFC-134a being particularly preferable in view of the environmental problems.

The refrigerator oils of the present invention are very excellent in compatibility with the hydrogen-containing halogenocarbons as compared with the heretofore known refrigerator oils. Further, the refrigerator oils of the present invention are excellent ones which have not only high compatibility with the hydrogen-containing halogenocarbons and high electrical insulating property but also high lubricity and low hygroscopicity.

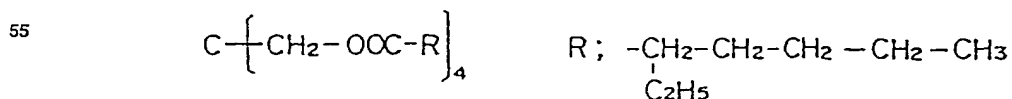
The refrigerator oils of the present invention may particularly preferably be used in refrigerators, air-conditioners, dehumidifiers, cold-storage chests, freezers, freeze and refrigeration warehouses, automatic vending machines, showcases, cooling units in chemical plants, and the like which have a reciprocating or rotary compressor. Further, the above refrigerator oils may also preferably be used in refrigerators having a centrifugal compressor.

This invention will be better understood by the following Examples and Comparative Examples.

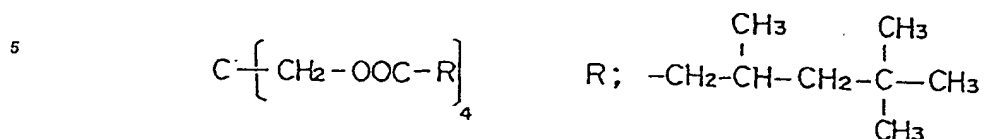
#### Examples 1-23 and Comparative Examples 1-10

Refrigerator oils of this invention and comparative refrigerator oils of a conventional type are illustrated as follows:

[Example 1] A tetraester of pentaerythritol (1 mol) and 2-ethylhexanoic acid (4 mol):

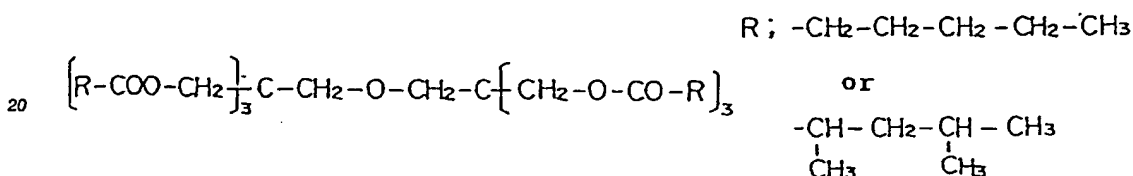


[Example 2] A tetraester of pentaerythritol (1 mol) and 3,5,5-trimethylhexanoic acid (4 mol):

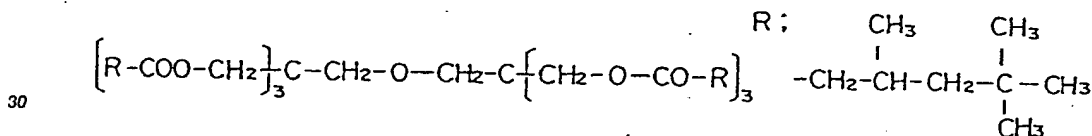


[Example 3] A tetraester of pentaerythritol (1 mol), 2-ethylhexanoic acid (2 mol) and 3,5,5-trimethylhexanoic acid (2 mol)

[Example 4] A hexaester of dipentaerythritol (1 mol), n-hexanoic acid (3 mol) and 2,4-dimethylpentanoic acid (3 mol):

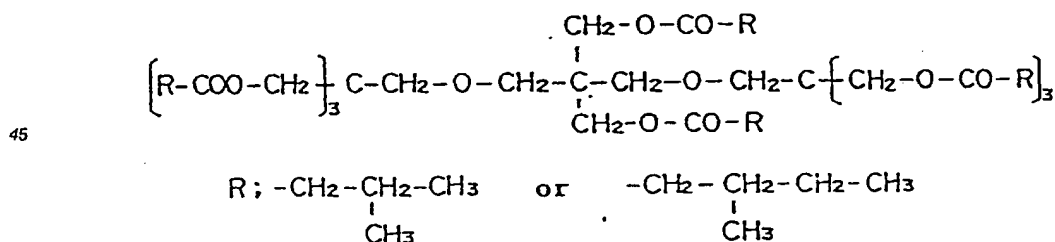


[Example 5] A hexaester of dipentaerythritol (1 mol) and 3,5,5-trimethylhexanoic acid (6 mol):

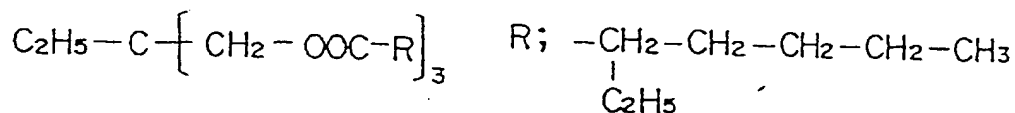


[Example 6] A mixture of 50 parts by weight of the same ester as in Example 1 and 50 parts by weight of the same ester as in Example 5

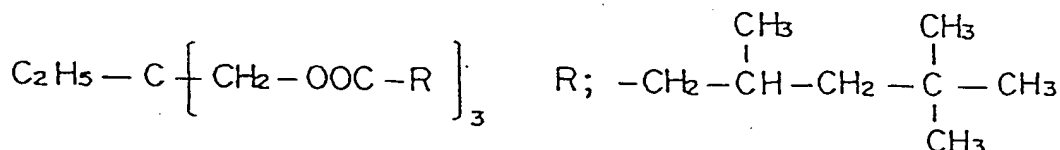
[Example 7] A mixture of 30 parts by weight of the same ester as in Example 2, 40 parts by weight of the same ester as in Example 5 and 30 parts by weight of the following octaester of tripentaerythritol (1 mol), 3-methylbutanoic acid (4 mol) and 3-methylpentanoic acid (4 mol):



[Example 8] A triester of trimethylolpropane (1 mol) and 2-ethylhexanoic acid (3 mol):

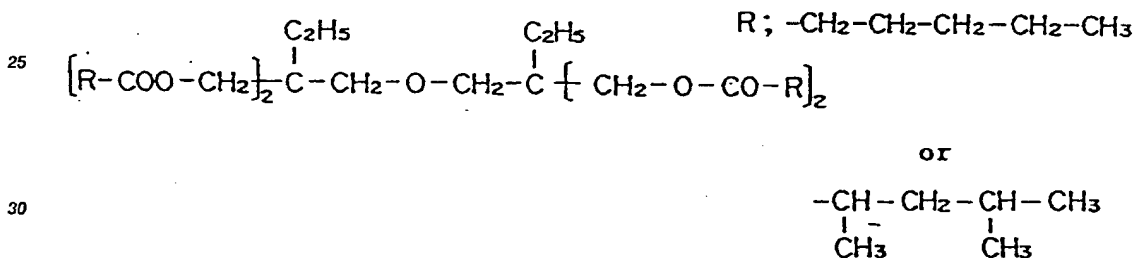


[Example 9] A triester of trimethylolpropane (1 mol) and 3,5,5-trimethylhexanoic acid (3 mol):

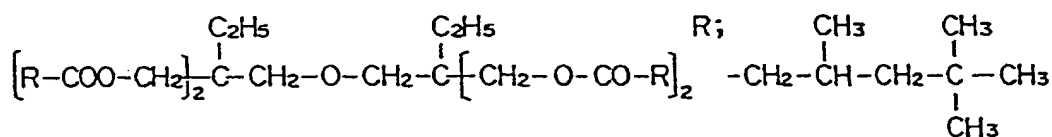


[Example 10] A triester of trimethylolpropane (1 mol), 2-ethylhexanoic acid (1.5 mol) and 3,5,5-trimethylhexanoic acid (1.5 mol)

[Example 11] A tetraester of di-(trimethylolpropane) (1 mol), n-hexanoic acid (2 mol) and 2,4-dimethylpentanoic acid (2 mol):

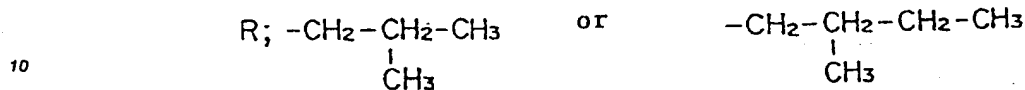
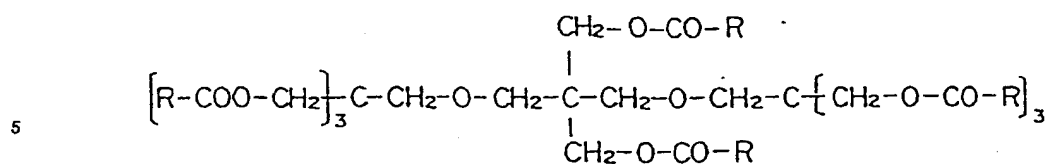


[Example 12] A tetraester of di-(trimethylolpropane) (1 mol) and 3,5,5-trimethylhexanoic acid (4 mol):

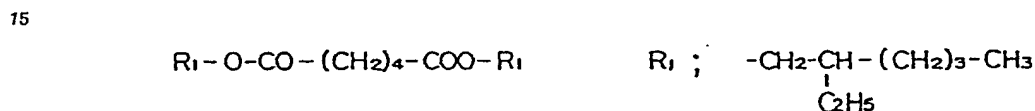


[Example 13] A mixture of 50 parts by weight of the same ester as in Example 8 and 50 parts by weight of the same ester as in Example 12

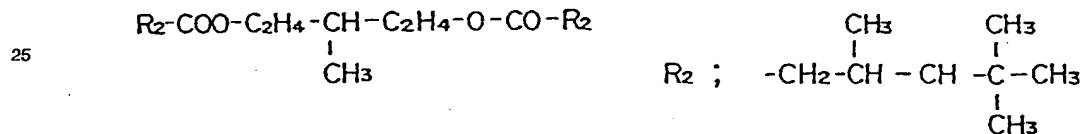
[Example 14] A mixture of 30 parts by weight of the same ester as in Example 9, 40 parts by weight of the same ester as in Example 12 and 30 parts by weight of the following octaester of tripentaerythritol (1 mol), 3-methylbutanoic acid (4 mol) and 3-methylpentanoic acid (4 mol):



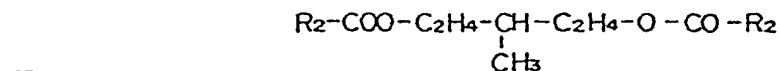
[Example 15] A diester of adipic acid and 2-ethyl-1-hexanol:



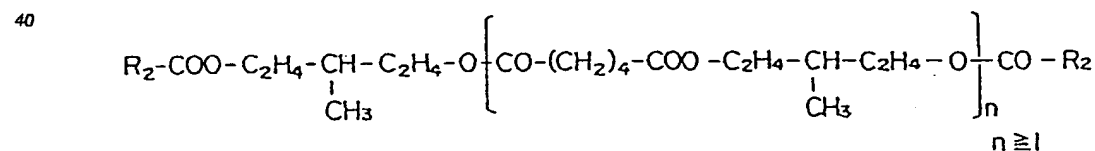
[Example 16] A diester of 3-methyl-1,5-pentanediol and 3,5,5-trimethylhexanoic acid:



[Example 17] A complex ester having an average molecular weight of about 500 of 3-methyl-1,5-pentanediol, adipic acid and 3,5,5-trimethylhexanoic acid:



and



R<sub>2</sub>; the same as in Example 16

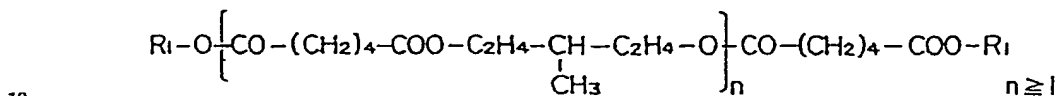
[Example 18] A complex ester having the same composition as that in Example 17 except for having an average molecular weight of about 700

[Example 19] A complex ester having an average molecular weight of about 520 of 3-methyl-1,5-pentanediol, adipic acid and 2-ethyl-1-hexanol:



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and



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[Example 20] A complex ester having the same composition as that in Example 19 except for having an average molecular weight of about 750

15 [Example 21] A mixture of the same complex ester as in Example 17 and 1.0 wt.% of a phosphoric ester type wear inhibitor

[Example 22] An ester having an average molecular weight of 690 and a kinematic viscosity at 100 °C of 9.6 cSt, obtained by the reaction of 2 moles of trimethylolpropane with 1 mole of succinic acid, 1 mole of glutaric acid, 2 moles of iso-pentanoic acid and 2 moles of iso-hexanoic acid

20 [Example 23] An ester having an average molecular weight of 880 and a kinematic viscosity at 100 °C of 12.7 cSt, obtained by the reaction of 2 moles of pentaerythritol with 1 mole of succinic acid, 1 mole of glutaric acid, 3 moles of iso-pentanoic acid and 3 moles of iso-hexanoic acid

[Comparative Example 1] A naphthenic mineral oil

[Comparative Example 2] A branched-chain type alkylbenzene (average molecular weight: about 300)

25 [Comparative Example 3] Polyoxypropylene glycol monobutyl ether (average molecular weight: about 500)

[Comparative Example 4] Polyoxypropylene glycol monobutyl ether (average molecular weight: about 1000)

[Comparative Example 5] Polyoxypropylene glycol (average molecular weight: about 700)

[Comparative Example 6] Polyoxypropylene glycol (average molecular weight: about 2000)

30 [Comparative Example 7] A tetraester of pentaerythritol (1 mol) and n-nonanoic acid (4 mol)

[Comparative Example 8] A tetraester of pentaerythritol (1 mol) and coconut oil

[Comparative Example 9] A triester of trimethylolpropane (1 mol) and n-nonanoic acid (3 mol)

[Comparative Example 10] A triester of trimethylolpropane (1 mol) and coconut oil

35 The base oils of Examples 1-23 for the refrigerator oils of the present invention were evaluated for their performances that are their compatibility with HFC-134a, insulating property, wear resistance and hygroscopicity by the following respective test methods. For comparison, the mineral oil, the alkylbenzene, the polyglycols and the esters of the Comparative Examples which have heretofore been used as refrigerator oils were evaluated in the same manner as in Examples 1-23.

40

(Miscibility with HFC-134a)

0.2 g of the test oil of each of the Examples and the Comparative Examples and 1.8 g of the refrigerant (HFC-134a) were sealed in a glass tube having an inner diameter of 6 mm and a length of 220 mm. This glass tube was then placed in a thermostat maintained at a predetermined low temperature or high temperature to observe whether the refrigerant and the test oil were miscible with each other, separated from each other or made white-turbid.

45

(Insulating property)

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The test oils were each measured for specific volume resistivity at 25 °C in accordance with JIS C 2101.

(FALEX wear test)

55

The test oils were each applied to a test journal for measuring the amount of the test journal worn by having the journal run in at a test oil temperature of 100 °C under a load of 150 lb for 1 minute and then running it under a load of 250 lb for 2 hours in accordance with ASTM D 2670.

(Hygroscopicity)

Thirty grams (30 g) of each of the test oils were placed in a 300-ml beaker, allowed to stand for 7 days in an air-conditioned bath maintained at a temperature of 60 °C and a humidity of 30 % and then measured for water content by the Karl-Fischer method.

Table 1

Example	Kinematic viscosity @ 100°C (cSt)	Miscibility with HFC-134a Miscible temperature range (°C)	Resistivity @25°C (Ω·cm)	FALEX test Amount of journal worn (mg)	Hygroscopicity 60°C, 30% (%)
Example1	6.2	-27 - CT *	$4.1 \times 10^{14}$	27	0.19
Example2	11.5	-32 - CT *	$3.8 \times 10^{14}$	25	0.19
Example3	8.1	-28 - CT *	$4.0 \times 10^{14}$	26	0.19
Example4	10.0	<-70 - CT *	$2.8 \times 10^{14}$	22	0.17
Example5	28.2	- 7 - 75	$3.0 \times 10^{14}$	19	0.15
Example6	11.8	-25 - 90	$3.2 \times 10^{14}$	22	0.16
Example7	32.3	- 5 - 86	$2.4 \times 10^{14}$	18	0.13
Example8	3.8	-35 - CT *	$4.5 \times 10^{14}$	29	0.20
Example9	9.5	-45 - CT *	$3.9 \times 10^{14}$	26	0.20
Example10	7.2	-37 - CT *	$4.1 \times 10^{14}$	26	0.20
Example11	9.1	<-70 - CT *	$2.5 \times 10^{14}$	24	0.18
Example12	22.1	-11 - 75	$2.8 \times 10^{14}$	20	0.14
Example13	9.8	-20 - 0	$3.0 \times 10^{14}$	24	0.16
Example14	31.9	- 7 - 88	$2.5 \times 10^{14}$	19	0.14
Example15	2.3	-24 - CT *	$4.2 \times 10^{14}$	24	0.22
Example16	3.1	-30 - CT *	$3.9 \times 10^{14}$	22	0.20
Example17	5.5	-55 - CT *	$2.5 \times 10^{14}$	15	0.18
Example18	10.3	-34 - 95	$1.4 \times 10^{14}$	12	0.11
Example19	6.2	-47 - CT *	$2.2 \times 10^{14}$	14	0.16
Example20	11.6	-30 - 92	$1.1 \times 10^{14}$	10	0.10
Example21	5.5	-55 - CT *	$2.3 \times 10^{14}$	2	0.18
Example22	9.6	<-35 - 76	$2.8 \times 10^{14}$	20	0.18
Example23	12.7	<-30 - 68	$1.2 \times 10^{14}$	17	0.12

\*: CT; Critical temperature of HFC-134a (102°C)

Table 2

Com- parative Example	Kinematic viscosity @ 100°C (cSt)	Miscibility with HFC-134a Miscible temperature range (°C)	Resistivity @25°C (Ω·cm)	FALEX test Amount of journal worn (mg)	Hygro- scopicity 60°C, 30% (%)
Com.Ex.1	5.1	Immiscible	$3.8 \times 10^{15}$	23	0.01
Com.Ex.2	4.8	Immiscible	$6.4 \times 10^{15}$	25	0.01
Com.Ex.3	4.9	<-70 - 97	$1.1 \times 10^{11}$	40	1.31
Com.Ex.4	10.8	<-70 - 56	$1.2 \times 10^{11}$	35	1.01
Com.Ex.5	10.6	<-70 - 64	$5.6 \times 10^{10}$	38	2.30
Com.Ex.6	22.4	-51 - 32	$4.8 \times 10^{10}$	30	1.81
Com.Ex.7	6.1	Immiscible	-	-	-
Com.Ex.8	4.9	Immiscible	-	-	-
Com.Ex.9	5.4	Immiscible	-	-	-
Com.Ex.10	4.4	Immiscible	-	-	-

It is apparent from the results indicated in Table 1 that the refrigerator oils (Examples 1-23) of the present invention are very excellent in miscibility with a refrigerant, HFC-134a, as compared with those of Comparative Examples 1-2 and 7-10.

As is apparent from the results indicated in Table 2, the esters of Comparative Examples 7 and 9 in which the alkyl groups of the acid moiety are all straight-chain ones are inferior in miscibility with the refrigerant. Further, Table 2 shows that the esters of Comparative Examples 8 and 10 which have heretofore been used as lubricating oils, refrigerator oils and the like and in which the acid moiety is originated from a natural fat or oil, are also inferior in miscibility with the refrigerant.

Table 2 further shows that the ethers of Comparative Examples 3 and 4 as well as the polyoxypropylene glycols of Comparative Examples 5 and 6 are excellent in miscibility with the refrigerant, but these ethers and glycols are inferior in insulating property thereby rendering them unusable for hermetic type compressors. Still further, Tables 1 and 2 show that the ethers and glycols of Comparative Examples 3-6 have hygroscopicity 5-10 times that of the refrigerator oils of Examples 1-23 and are also inferior in electrical insulating property, ice choke, wear resistance, stability and the like to the refrigerator oils of the Examples.

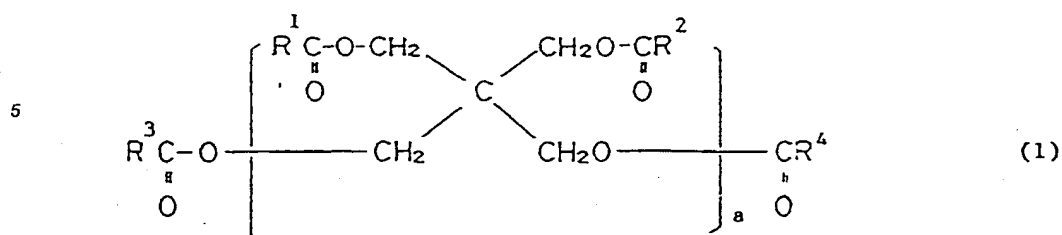
The FALEX wear test shows that the refrigerator oils of Examples 1-23 are at least equal in wear resistance to those of Comparative Examples 3-6.

(Effect of the Invention)

As is apparent from the above comparative experiments, lubricating oils for compressors of refrigerators of the present invention are suitable for use in refrigerators using therein a hydrogen-containing halogenocarbon as a refrigerant and are excellent in electrical insulating property as well as wear resistance and nonhygroscopicity.

#### Claims

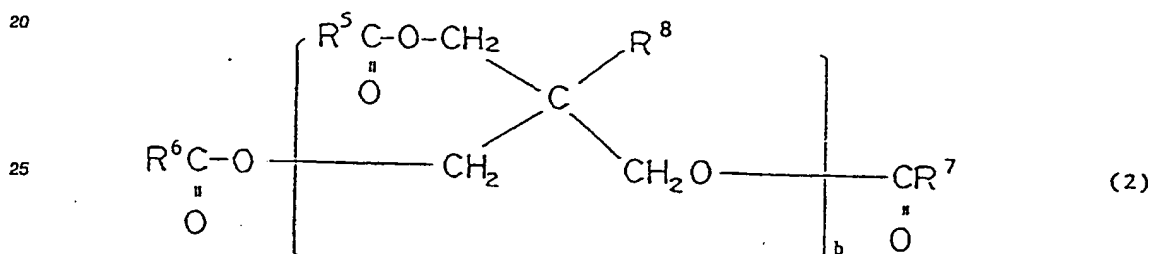
1. A refrigerator oil for use in compressors using therein a hydrogen-containing halogenocarbon as a refrigerant, comprising as a base oil at least one kind of an ester selected from the group consisting of:  
[I] a pentaerythritol ester represented by the general formula (I)



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wherein R<sup>1</sup>-R<sup>4</sup> may be identical with, or different from, each other and are each a group selected from the group consisting of straight-chain alkyl groups having 3-11 carbon atoms, branched-chain alkyl groups having 3-15 carbon atoms and cycloalkyl groups having 6-12 carbon atoms, the straight-chain alkyl groups being present in a ratio of not more than 60 % of the total alkyl groups, and a is an integer of 1-3;

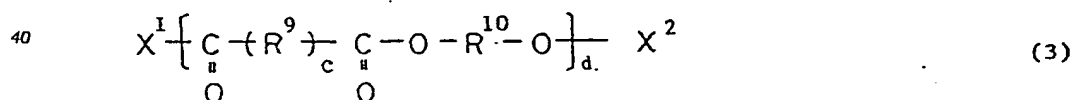
[II] a polyol ester represented by the general formula (2)



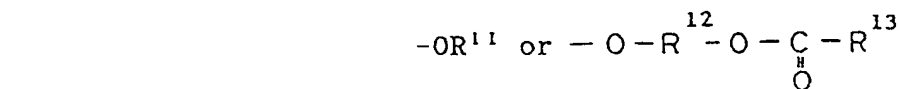
**30**

therein R<sup>5</sup>-R<sup>7</sup> may be identical with, or different from, each other and are each a group selected from the group consisting of straight-chain alkyl groups having 3-11 carbon atoms, branched-chain alkyl groups having 3-15 carbon atoms and cycloalkyl groups having 6-12 carbon atoms, the straight-chain alkyl groups being present in a ratio of not more than 60 % of the total alkyl groups, R<sup>8</sup> is a group selected from the group consisting of methyl, ethyl and propyl groups, and b is an integer of 1-3;

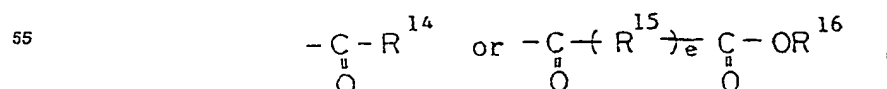
[III] an ester represented by the general formula (3)



45 wherein  $X^1$  is a group represented by the general formula



$X^2$  is a group represented by the general formula





R<sup>9</sup> and R<sup>15</sup> are each a divalent hydrocarbon group having 1-8 carbon atoms, R<sup>10</sup> and R<sup>12</sup> are each a divalent saturated hydrocarbon group having 2-16 carbon atoms, R<sup>11</sup> and R<sup>16</sup> are each an alkyl group having 1-15 carbon atoms, R<sup>13</sup> and R<sup>14</sup> are each an alkyl group having 1-14 carbon atoms, c and e are each an integer of 0 or 1 and d is an integer of 0-30;

5 and

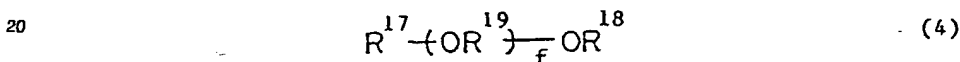
[IV] a polyol ester obtained by the synthesis of, as raw materials, (a) a neopentyl type polyhydric alcohol having 5-6 carbon atoms and 3-4 hydroxyl groups, (b) a monocarboxylic acid and (c) a dicarboxylic acid.

10 2. A refrigerator oil according to claim 1, wherein the hydrogen-containing halogenocarbon is a chlorine-free type halogenocarbon.

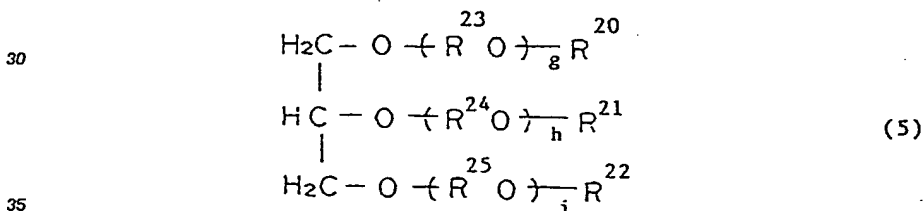
3. A refrigerator oil according to claim 1, wherein the base oil is a mixture oil of:

15 (A) at least one kind of an ester selected from the group consisting of the pentaerythritol ester [I], the polyol ester [II], the ester [III] and the polyol ester [IV] and

(B) at least one kind of an oil selected from the group consisting of a polyoxyalkylene glycol or an ether thereof represented by the general formula (4)



25 wherein R<sup>17</sup> and R<sup>18</sup> are each a hydrogen atom or an alkyl group having 1-18 carbon atoms, R<sup>19</sup> is an alkylene group having 2-4 carbon atoms, and f is an integer of 5-70, and a polyoxyalkylene glycol glycerol ether represented by the general formula (5)



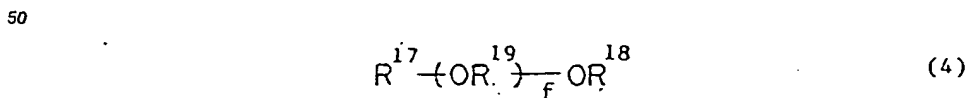
35 wherein R<sup>20</sup>-R<sup>22</sup> are each a hydrogen atom or an alkyl group having 1-18 carbon atoms, R<sup>23</sup>-R<sup>25</sup> are each an alkylene group having 2-4 carbon atoms, and g-i are each an integer of 5-7.

40 4. A refrigerator oil according to claim 3, wherein the ester (A) is comprised in an amount of more than 50 % by weight based on the total amount of the refrigerator oil.

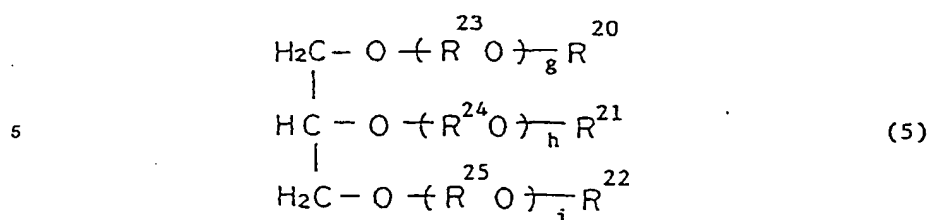
5. A refrigerator oil according to claim 2, wherein the base oil is a mixture oil of:

45 (A) at least one kind of an ester selected from the group consisting of the pentaerythritol ester [I], the polyol ester [II], the ester [III] and the polyol ester [IV] and

(B) at least one kind of an oil selected from the group consisting of a polyoxyalkylene glycol or an ether thereof represented by the general formula (4)



55 wherein R<sup>17</sup> and R<sup>18</sup> are each a hydrogen atom or an alkyl group having 1-18 carbon atoms, R<sup>19</sup> is an alkylene group having 2-4 carbon atoms, and f is an integer of 5-70, and a polyoxyalkylene glycol glycerol ether represented by the general formula (5)



wherein  $\text{R}^{20}$ - $\text{R}^{22}$  are each a hydrogen atom or an alkyl group having 1-18 carbon atoms,  $\text{R}^{23}$ - $\text{R}^{25}$  are each an alkylene group having 2-4 carbon atoms, and  $g$ - $i$  are each an integer of 5-7.

6. A refrigerator oil according to claim 5, wherein the ester (A) is comprised in an amount of more than 50 % by weight based on the total amount of the refrigerator oil.
7. A refrigerator oil according to claim 1, further comprising at least one kind of a phosphorous compound selected from the group consisting of phosphoric esters, acid phosphoric esters, amine salts of acid phosphoric esters, chlorinated phosphoric esters and phosphorous esters, in an amount of 0.1-5.0 % by weight based on the total amount of the refrigerator oil.
8. A refrigerator oil according to claim 2, further comprising at least one kind of a phosphorous compound selected from the group consisting of phosphoric esters, acid phosphoric esters, amine salts of acid phosphoric esters, chlorinated phosphoric esters and phosphorous esters, in an amount of 0.1-5.0 % by weight based on the total amount of the refrigerator oil.
9. A refrigerator oil according to claim 3, further comprising at least one kind of a phosphorous compound selected from the group consisting of phosphoric esters, acid phosphoric esters, amine salts of acid phosphoric esters, chlorinated phosphoric esters and phosphorous esters, in an amount of 0.1-5.0 % by weight based on the total amount of the refrigerator oil.
10. A refrigerator oil according to claim 4, further comprising at least one kind of a phosphorous compound selected from the group consisting of phosphoric esters, acid phosphoric esters, amine salts of acid phosphoric esters, chlorinated phosphoric esters and phosphorous esters, in an amount of 0.1-5.0 % by weight based on the total amount of the refrigerator oil.
11. A refrigerator oil according to claim 5, further comprising at least one kind of a phosphorous compound selected from the group consisting of phosphoric esters, acid phosphoric esters, amine salts of acid phosphoric esters, chlorinated phosphoric esters and phosphorous esters, in an amount of 0.1-5.0 % by weight based on the total amount of the refrigerator oil.
12. A refrigerator oil according to claim 6, further comprising at least one kind of a phosphorous compound selected from the group consisting of phosphoric esters, acid phosphoric esters, amine salts of acid phosphoric esters, chlorinated phosphoric esters and phosphorous esters, in an amount of 0.1-5.0 % by weight based on the total amount of the refrigerator oil.
13. A refrigerator oil according to claim 1, further comprising at least one kind of an epoxy compound selected from the group consisting of phenylglycidyl ether type epoxy compounds, glycidyl ester type epoxy compounds, epoxidized fatty acid monoesters and epoxidized vegetable oils, in an amount of 0.1-5.0 % by weight based on the total amount of the refrigerator oil.
14. A refrigerator oil according to claim 2, further comprising at least one kind of an epoxy compound selected from the group consisting of phenylglycidyl ether type epoxy compounds, glycidyl ester type epoxy compounds, epoxidized fatty acid monoesters and epoxidized vegetable oils, in an amount of 0.1-5.0 % by weight based on the total amount of the refrigerator oil.
15. A refrigerator oil according to claim 3, further comprising at least one kind of an epoxy compound selected from the group consisting of phenylglycidyl ether type epoxy compounds, glycidyl ester type

epoxy compounds, epoxidized fatty acid monoesters and epoxidized vegetable oils, in an amount of 0.1-5.0 % by weight based on the total amount of the refrigerator oil.

16. A refrigerator oil according to claim 4, further comprising at least one kind of an epoxy compound selected from the group consisting of phenylglycidyl ether type epoxy compounds, glycidyl ester type epoxy compounds, epoxidized fatty acid monoesters and epoxidized vegetable oils, in an amount of 0.1-5.0 % by weight based on the total amount of the refrigerator oil.
17. A refrigerator oil according to claim 5, further comprising at least one kind of an epoxy compound selected from the group consisting of phenylglycidyl ether type epoxy compounds, glycidyl ester type epoxy compounds, epoxidized fatty acid monoesters and epoxidized vegetable oils, in an amount of 0.1-5.0 % by weight based on the total amount of the refrigerator oil.
18. A refrigerator oil according to claim 6, further comprising at least one kind of an epoxy compound selected from the group consisting of phenylglycidyl ether type epoxy compounds, glycidyl ester type epoxy compounds, epoxidized fatty acid monoesters and epoxidized vegetable oils, in an amount of 0.1-5.0 % by weight based on the total amount of the refrigerator oil.
19. A refrigerator oil according to claim 7, further comprising at least one kind of an epoxy compound selected from the group consisting of phenylglycidyl ether type epoxy compounds, glycidyl ester type epoxy compounds, epoxidized fatty acid monoesters and epoxidized vegetable oils, in an amount of 0.1-5.0 % by weight based on the total amount of the refrigerator oil.
20. A refrigerator oil according to claim 8, further comprising at least one kind of an epoxy compound selected from the group consisting of phenylglycidyl ether type epoxy compounds, glycidyl ester type epoxy compounds, epoxidized fatty acid monoesters and epoxidized vegetable oils, in an amount of 0.1-5.0 % by weight based on the total amount of the refrigerator oil.
21. A refrigerator oil according to claim 9, further comprising at least one kind of an epoxy compound selected from the group consisting of phenylglycidyl ether type epoxy compounds, glycidyl ester type epoxy compounds, epoxidized fatty acid monoesters and epoxidized vegetable oils, in an amount of 0.1-5.0 % by weight based on the total amount of the refrigerator oil.
22. A refrigerator oil according to claim 10, further comprising at least one kind of an epoxy compound selected from the group consisting of phenylglycidyl ether type epoxy compounds, glycidyl ester type epoxy compounds, epoxidized fatty acid monoesters and epoxidized vegetable oils, in an amount of 0.1-5.0 % by weight based on the total amount of the refrigerator oil.
23. A refrigerator oil according to claim 11, further comprising at least one kind of an epoxy compound selected from the group consisting of phenylglycidyl ether type epoxy compounds, glycidyl ester type epoxy compounds, epoxidized fatty acid monoesters and epoxidized vegetable oils, in an amount of 0.1-5.0 % by weight based on the total amount of the refrigerator oil.
24. A refrigerator oil according to claim 12, further comprising at least one kind of an epoxy compound selected from the group consisting of phenylglycidyl ether type epoxy compounds, glycidyl ester type epoxy compounds, epoxidized fatty acid monoesters and epoxidized vegetable oils, in an amount of 0.1-5.0 % by weight based on the total amount of the refrigerator oil.



European Patent  
Office

# EUROPEAN SEARCH REPORT

Application Number

EP 90 12 5500

DOCUMENTS CONSIDERED TO BE RELEVANT			
Category	Citation of document with indication, where appropriate, of relevant passages	Relevant to claim	CLASSIFICATION OF THE APPLICATION (Int. Cl. 5)
X	GB-A-2 216 541 (ICI) * Page 2, lines 8-16; page 2, line 33 - page 3, line 2; page 3, lines 15-18 *	1,2	C 10 M 105/36 C 10 M 105/38 C 10 M 105/42
A	---	3,5	C 10 M 111/04 C 10 M 169/04
D,X	US-A-4 851 144 (P.W. McGRAW) * Column 2, lines 47-50; column 3, lines 4-9,50-52 *	1,2,3,5 -7,8,9, 11	C 09 K 5/04. // (C 10 M 111/04 C 10 M 105:36 C 10 M 105:38 C 10 M 105:42 C 10 M 107:34 ) (C 10 M 169/04 C 10 M 105:36 C 10 M 105:38 C 10 M 105:42 C 10 M 107:34 C 10 M 129:18 -/-
A	---	4,6	
X	DATABASE WPIL, accession no. 87-041021 [06], Derwent Publications Ltd, London, GB; & JP-A-62 000 592 (NIPPON OILS) 06-01-1987 * Abstract *	1	
X	DATABASE WPIL, accession no. 84-267230 [43], Derwent Publications Ltd, London, GB; & JP-A-59 164 393 (NIPPON OILS) 17-09-1984 * Abstract *	1	
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		-/-	
The present search report has been drawn up for all claims			
Place of search THE HAGUE		Date of completion of the search 04-03-1991	Examiner HILGENGA K.J.
<p>CATEGORY OF CITED DOCUMENTS</p> <p>X : particularly relevant if taken alone Y : particularly relevant if combined with another document of the same category A : technological background O : non-written disclosure P : intermediate document</p> <p>T : theory or principle underlying the invention E : earlier patent document, but published on, or after the filing date D : document cited in the application L : document cited for other reasons ----- &amp; : member of the same patent family, corresponding document</p>			



European Patent  
Office

# EUROPEAN SEARCH REPORT

Page 2

Application Number

EP 90 12 5500

DOCUMENTS CONSIDERED TO BE RELEVANT			
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A	DATABASE WPIL, accession no. 88-033105 [05], Derwent Publications Ltd, London, GB; & JP-A-62 292 895 (NIPPON OIL) 19-12-1987 * Abstract *	1, 3, 13, 15	C 10 M 137:02 ) C 10 N 40:30
P, X	WO-A-9 012 849 (THE LUBRIZOL CORP.) * Page 9, line 7; page 10, lines 12-25; page 12, lines 1-3; page 14, lines 1-7; page 14, lines 21-32; page 26, lines 18-20 *	1, 2, 7, 8	
X	US-A-2 807 155 (V.A. WILLAIMITIS) * Column 3, lines 29-30; claim 4 *	1	
			TECHNICAL FIELDS SEARCHED (Int. Cl.5)
The present search report has been drawn up for all claims			
Place of search THE HAGUE		Date of completion of the search 04-03-1991	Examiner HILGENA K.J.
<b>CATEGORY OF CITED DOCUMENTS</b>			
X : particularly relevant if taken alone Y : particularly relevant if combined with another document of the same category A : technological background O : non-written disclosure P : intermediate document		I : theory or principle underlying the invention E : earlier patent document, but published on, or after the filing date D : document cited in the application L : document cited for other reasons ..... & : member of the same patent family, corresponding document	

DERWENT-ACC-NO: 1991-268501  
DERWENT-WEEK: 199137  
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TITLE: Use of ester lubricating oils - in  
compressors for chlorine-free  
partially fluorinated hydrocarbon refrigerants

INVENTOR: CORNILS, B; HENRICI, R ; LAPPE, P ;  
PREISEGGER, E ; SPRINGER, H  
; WEBER, J

PATENT-ASSIGNEE: HOECHST AG[FARH]

PRIORITY-DATA: 1990DE-4006828 (March 5, 1990)

PATENT-FAMILY:

PUB-NO	PUB-DATE	
LANGUAGE	PAGES	MAIN-IPC
EP 445610 A	September 11, 1991	N/A
000	N/A	
DE 4006828 A	September 12, 1991	N/A
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DE 59101529 G	June 9, 1994	N/A
000	C10M 105/00	
EP 445610 B1	May 4, 1994	G
006	C10M 105/00	
ES 2056507 T3	October 1, 1994	N/A
000	C10M 105/00	
JP 04220497 A	August 11, 1992	N/A
008	C10M 105/44	

DESIGNATED-STATES: AT BE DE ES FR GB IT NL SE AT BE  
DE ES FR GB IT NL SE

CITED-DOCUMENTS: EP 218207; EP 406479 ; GB 1460665  
; GB 2216541 ; JP59164393

; JP61171799 ; US 4155861 ; US 4851144 ; WO 9012849  
; 02Jnl.Ref ; US 4166861

APPLICATION-DATA:

PUB-NO	APPL-DESCRIPTOR	APPL-NO
	APPL-DATE	
EP 445610A	N/A	
1991EP-0102680	February 23, 1991	
DE 4006828A	N/A	
1990DE-4006828	March 5, 1990	
DE59101529G	N/A	
1991DE-0501529	February 23, 1991	
DE59101529G	N/A	
1991EP-0102680	February 23, 1991	
DE59101529G	Based on	EP
445610	N/A	
EP 445610B1	N/A	
1991EP-0102680	February 23, 1991	
ES 2056507T3	N/A	
1991EP-0102680	February 23, 1991	
ES 2056507T3	Based on	EP
445610	N/A	
JP04220497A	N/A	
1991JP-0034478	February 28, 1991	

INT-CL (IPC): C09K005/04; C10M105/42 ;  
C10M105/44 ; C10N040/30 ;  
C10N040:30 ; C10M105/00 ; C10M105:42 ;  
C10M105:52

ABSTRACTED-PUB-NO: EP 445610A

BASIC-ABSTRACT: Certain esters (I) are used as  
lubricants for refrigerant  
compressors operating with Cl-free partially  
fluorinated hydrocarbon  
refrigerants.

(I) are esters derived from 'dicarboxylic acids and  
glycols or polyglycols and  
monocarboxylic acids or monoalcohols' (sic).

ADVANTAGE - (I) exhibit good miscibility with the refrigerants at temps. as low as -40 deg.C., have viscosities of 10-100 mm<sup>2</sup>/sec. at 40 deg.C., have good thermal stability in the absence of air and moisture, and are non-hygroscopic.

ABSTRACTED-PUB-NO: EP 445610B

EQUIVALENT-ABSTRACTS: The use of esters prepared from a dicarboxylic acid through partial esterification with a sub-stoichiometric amount of a glycol or polyglycol as well as through completion of the esterification of the acid with the necessary amount of a monoalcohol or prepared from a glycol or polyglycol through partial esterification with a sub-stoichiometric amount of a dicarboxylic acid as well as through completion of the esterification of the glycol or polyglycol with the necessary amount of a monocarboxylic acid as lubricants for refrigeration compressors utilising chlorine-free, partially fluorinated hydrocarbons as refrigerant s.

CHOSEN-DRAWING: Dwg.0/0 Dwg.0/0

TITLE-TERMS:

ESTER LUBRICATE OIL COMPRESSOR CHLORINE FREE  
FLUORINATED HYDROCARBON  
REFRIGERATE

DERWENT-CLASS: A97 E17 H07 J07

CPI-CODES: A05-E01D; A12-W02; E10-E04K; E10-G02G;  
E10-G02H; H07-A; J07-A04;

CHEMICAL-CODES:

Chemical Indexing M3 \*01\*  
Fragmentation Code



H581 H582 H583 H584 H589 J0 J014 J2 J273  
M210  
M211 M212 M213 M214 M215 M216 M220 M221 M222  
M223  
M224 M225 M226 M231 M232 M233 M262 M272 M281  
M282  
M312 M313 M314 M315 M316 M321 M322 M323 M331  
M332  
M333 M342 M382 M383 M391 M392 M393 M416 M620  
M781  
M903 M904 Q416 Q433 R023  
Markush Compounds  
199137-A7201-P 199137-A7202-P 199137-A7203-P  
199137-A7204-P

UNLINKED-DERWENT-REGISTRY-NUMBERS: 0930S; 1060S

POLYMER-MULTIPUNCH-CODES-AND-KEY-SERIALS:

Key Serials: 0004 0013 0037 0231 1279 1291 1319

1321 1323 1325 1327 3075 1329

1450 1452 1454 1594 1608 1999 2014 2177 2197 2198

2511 2564 2585 2707 2758

Multipunch Codes: 014 028 038 04- 075 143 144 147

155 157 160 161 162 169 170

171 172 173 198 200 207 208 231 239 31- 336 359 475

512 53& 575 583 589 638 644

720

SECONDARY-ACC-NO:

CPI Secondary Accession Numbers: C1991-116363